

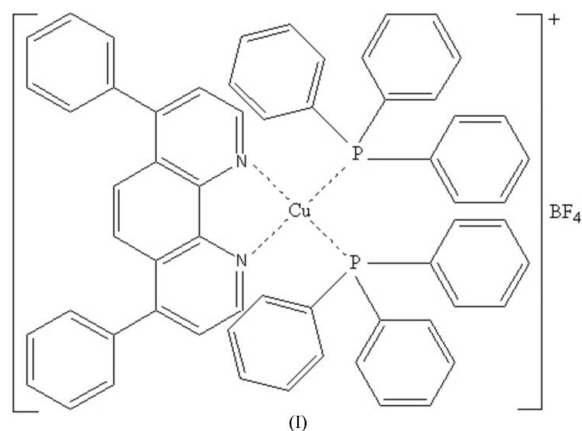
## (4,7-Diphenyl-1,10-phenanthroline)bis(triphenylphosphine)copper(I) tetrafluoroborate

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## Key indicators

Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$   
 $R$  factor = 0.045  
 $wR$  factor = 0.136  
Data-to-parameter ratio = 15.8For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The title compound,  $[\text{Cu}(\text{C}_{24}\text{H}_{16}\text{N}_2)(\text{C}_{18}\text{H}_{15}\text{P})_2]\text{BF}_4$ , consists of mononuclear  $\text{Cu}(\text{PPh}_3)_2(\text{NN})^+$  and  $\text{BF}_4^-$  ions (NN is 4,7-diphenyl-1,10-phenanthroline). The  $\text{Cu}^{\text{I}}$  atom is four-coordinated by two P atoms of  $\text{PPh}_3$  groups and two N donors of two NN ligands to form a tetrahedral coordination geometry.

## Comment

It is well known that the mixed-ligand complexes of the  $\text{Cu}(\text{PPh}_3)_2(\text{NN})^+$  cation, where  $\text{PPh}_3$  and NN denote triphenylphosphine and 4,7-diphenyl-1,10-phenanthroline, respectively, both exhibit strong phosphorescent emissions in the solid state at room temperature. These  $\text{Cu}^{\text{I}}$  complexes have been investigated for a long time, with regard to their preparation (Kirchhoff *et al.*, 1985), photophysical properties (Palmer *et al.*, 1987) and so on. However, reports on the crystal structures of complexes with 4,7-diphenyl-1,10-phenanthroline or its derivatives are rare (Chesnut *et al.*, 2001). We report here the crystal structure of a 4,7-diphenyl-1,10-phenanthroline  $\text{Cu}^{\text{I}}$  complex, (I).Received 4 November 2005  
Accepted 30 November 2005  
Online 10 December 2005The title compound,  $\text{Cu}(\text{PPh}_3)_2(\text{NN})^+\cdot\text{BF}_4^-$ , consists of mononuclear  $\text{Cu}(\text{PPh}_3)_2(\text{NN})^+$  and  $\text{BF}_4^-$  ions. The  $\text{Cu}^{\text{I}}$  atom is four-coordinated by two P atoms of distinct  $\text{PPh}_3$  groups and two N donors of derivatives of 1,10-phenanthroline, to form a tetrahedral coordination geometry (Fig. 1). Selected geometric parameters are given in given in Table 1.

## Experimental

Complex (I) was synthesized according to the method of Cuttell *et al.* (2002). A mixture of  $[\text{Cu}(\text{NCCH}_3)_4]\text{BF}_4$  (310.0 mg, 1.0 mmol) and triphenylphosphine (524.6 mg, 2.0 mmol) in  $\text{CH}_2\text{Cl}_2$  (30 ml) was stirred at room temperature for 1 h, and then 4,7-diphenyl-1,10-phenanthroline (332.4 mg, 1.0 mmol) was added to the mixture. The

reaction mixture was stirred for an additional 1 h and filtered, and the clear yellow filtrate was concentrated to ca 12 ml. Acetonitrile (about 10 ml) was added, and the vapor diffusion of diethyl ether into the resulting solution gave yellow crystals of the complex (80% yield).

Crystal data

[Cu(C<sub>24</sub>H<sub>16</sub>N<sub>2</sub>)(C<sub>18</sub>H<sub>15</sub>P)<sub>2</sub>]BF<sub>4</sub>  
*M<sub>r</sub>* = 1007.28  
 Monoclinic, *P*2<sub>1</sub>/*n*  
*a* = 20.5552 (15) Å  
*b* = 11.9349 (9) Å  
*c* = 21.2725 (15) Å  
 $\beta$  = 103.870 (1)°  
*V* = 5066.5 (6) Å<sup>3</sup>  
*Z* = 4  
*D<sub>x</sub>* = 1.321 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 Cell parameters from 5088 reflections  
 $\theta$  = 2.3–25.9°  
 $\mu$  = 0.55 mm<sup>-1</sup>  
*T* = 293 (2) K  
 Plate, yellow  
 0.40 × 0.29 × 0.06 mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 1998)  
*T<sub>min</sub>* = 0.845, *T<sub>max</sub>* = 0.966  
 27936 measured reflections  
 9962 independent reflections  
 7512 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.028  
 $\theta_{max}$  = 26.0°  
*h* = -24 → 25  
*k* = -14 → 11  
*l* = -26 → 21

Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.045  
*wR* (*F*<sup>2</sup>) = 0.136  
*S* = 1.05  
 9962 reflections  
 631 parameters  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0841P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 ( $\Delta/\sigma$ )<sub>max</sub> = 0.001  
 $\Delta\rho_{max}$  = 0.70 e Å<sup>-3</sup>  
 $\Delta\rho_{min}$  = -0.36 e Å<sup>-3</sup>

Table 1

Selected geometric parameters (Å, °).

Cu–N2	2.0782 (19)	Cu–P2	2.2271 (6)
Cu–N1	2.0802 (17)	Cu–P1	2.2796 (7)
N2–Cu–N1	80.14 (7)	N2–Cu–P1	108.34 (5)
N2–Cu–P2	119.88 (5)	N1–Cu–P1	101.75 (5)
N1–Cu–P2	122.61 (5)	P2–Cu–P1	117.50 (2)

H atoms were included in calculated positions and treated as riding atoms, with C–H = 0.93 Å and *U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(C).

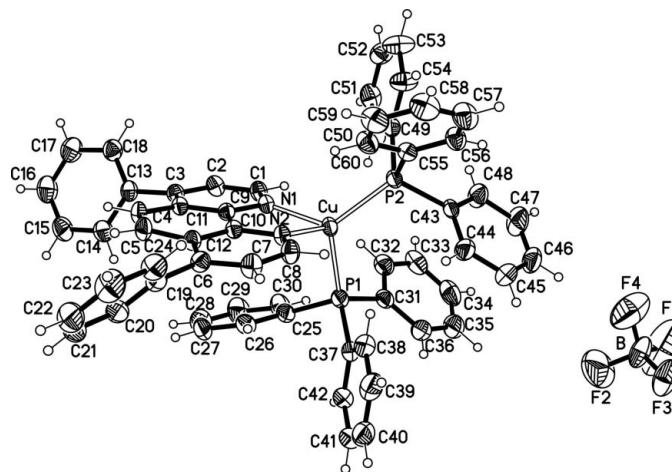


Figure 1  
 The molecular structure of the title compound, with 30% probability displacement ellipsoids.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1998); software used to prepare material for publication: SHELXTL.

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